

7.4. Exponential Growth and Decay

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Exponential Growth, for example, population growth...

$p \equiv$ population

$t \equiv$ time

$p_0 \equiv$ population at time $t=0$

The rate of increase of the population is proportional to the current population \Rightarrow

$$\frac{dp}{dt} = kp$$

$k \equiv$ proportionality constant, $k > 0$.

$\frac{dp}{p} = k dt$. Integrate from the initial state to some later state \Rightarrow

$$\int_{p_0}^p \frac{dz}{z} = k \int_0^t dx, \quad [\ln z]_{p_0}^p = k [x]_0^t, \quad \ln\left(\frac{p}{p_0}\right) = kt, \quad \frac{p}{p_0} = e^{kt}$$

$$p = p_0 e^{kt}$$

Example #1, In 1960, the population of France was 46.62 million people. In 2020, it was 67.39 million. Estimate the population of France in 2035.

Solution

$$p = p_0 e^{kt}, \quad p = 46.62 e^{kt} \quad (1960 \rightarrow t=0),$$

$$2020 \rightarrow t=60 \quad 67.39 = 46.62 e^{60k}, \quad e^{60k} = \frac{67.39}{46.62}, \quad 60k = \ln\left(\frac{67.39}{46.62}\right),$$

$$k = \frac{1}{60} \ln\left(\frac{67.39}{46.62}\right) = 0.00614.../\text{yr}$$

$$2035 \rightarrow t=75, \quad p = 46.62 e^{75k} = 73.89 \text{ million} \rightarrow$$

7.4. Exponential Growth and Decay

ZoF2

Exponential Decay, for example, radioactive decay...

$C \equiv$ concentration of a radioactive isotope

$t \equiv$ time

$C_0 \equiv$ concentration at time $t=0$

The rate of decrease of the concentration is proportional to the current concentration \Rightarrow

$$\frac{dC}{dt} = -kC$$

$k \equiv$ proportionality constant, $k > 0$.

$\frac{dC}{C} = -k dt$. Integrate from the initial state to some later state \Rightarrow

$$\int_{C_0}^C \frac{dC}{C} = -k \int_0^t dt, \quad [\ln C]_{C_0}^C = -k [t]_0^t, \quad \ln\left(\frac{C}{C_0}\right) = -kt, \quad \frac{C}{C_0} = e^{-kt}$$

$$C = C_0 e^{-kt}$$

Example #2. A sample containing 100% of radioactive Barium-137 (Ba-137) will contain 80% of its original Ba-137 after 50 seconds. After how many seconds will the sample contain 40% of the Ba-137?

Solution:

$$C = C_0 e^{-kt}, \quad 0.8 C_0 = C_0 e^{-50k}, \quad -50k = \ln 0.8, \quad k = -\frac{\ln 0.8}{50} = 0.00446/\text{sec}$$

$$0.4 C_0 = C_0 e^{-kt}, \quad -kt = \ln 0.4, \quad t = -\frac{\ln 0.4}{k} = 205.31 \text{ sec} = 3:25$$